## **General Properties of Enzymes**

- Enzymes differ from ordinary chemical catalysts in reaction rate, reaction conditions, reaction specificity, and control.
- The unique physical and chemical properties of the active site limit an enzyme's activity to specific substrates and reactions.
- Some enzymes require metal ions or organic cofactors.

## **Catalytic Mechanisms**

- Amino acid side chains that can donate or accept protons can participate in chemical reactions as acid or base catalysts: <u>Acid-base catalysis</u>
- 2. Nucleophilic groups can catalyze reactions through the transient formation of covalent bonds with the substrate: <u>Covalent catalysis</u>
- 3. In metal ion catalysis, the unique electronic properties of the metal ion facilitate the reaction: <u>Metal ion catalysis</u>
- 4. Enzymes accelerate reactions by bringing reacting groups together and orienting them for reaction: <a href="https://example.com/Proximity and orientation effects">Proximity and orientation effects</a>
- Transition state stabilization can significantly lower the activation energy for a reaction: <u>Preferential binding of the transition state</u> <u>complex</u>

## Lysozyme

- Model building indicates that binding to lysozyme distorts the substrate sugar residue.
- Lysozyme's active site Asp and Glu residues promote substrate hydrolysis by acid—base catalysis, covalent catalysis, and stabilization of an oxonium ion transition state.

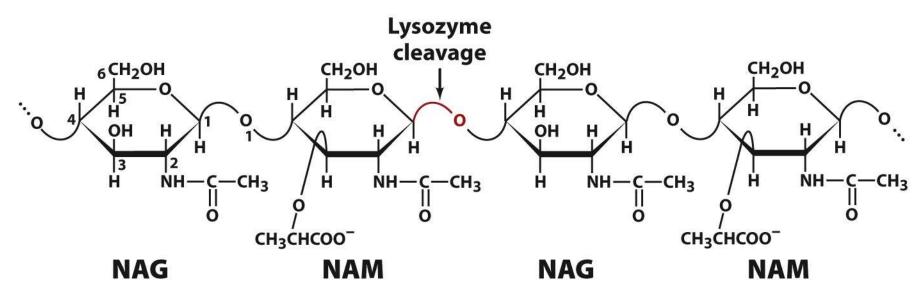


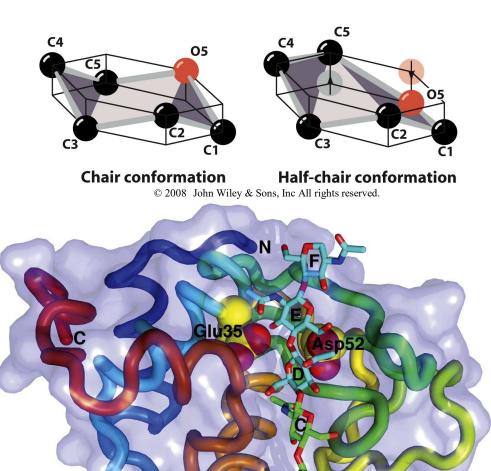
Figure 11-16 © 2013 John Wiley & Sons, Inc. All rights reserved.

## **Identification of Lysozyme Cleavage Site**

- Destroys bacterial cell walls (peptidoglycan)
- Hydrolyzing the  $\beta$  (1 $\rightarrow$ 4) glycosidic linkages from N-acetylmuramic acid (NAM) to N-acetylglucosamine (NAG)
- Also hydrolyzes  $\beta$  (1 $\rightarrow$ 4)-linked poly(NAG) (=chitin)
- Bactericidal agent or helps dispose of killed bacteria
- Hen egg white (HEW) lysozyme is the most studied.

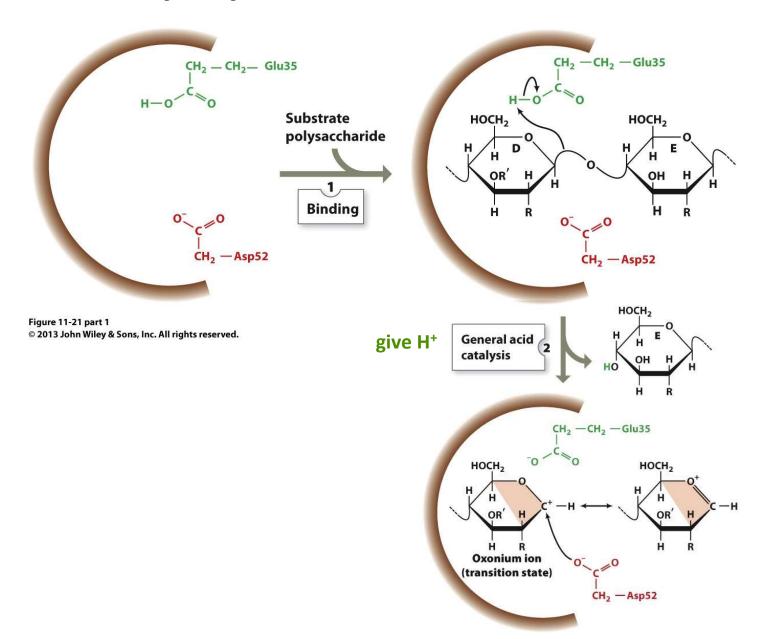
# HO NAG A Asp 101 CH<sub>2</sub> NAM Trp NAG D ring in NAM H--- 0= half-chair conformation Lysozyme cuts Gln Glu 35 Asn NAG Glu >=0

# **Lysozyme-Substrate Interactions**

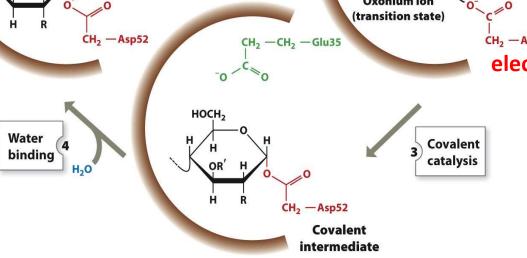


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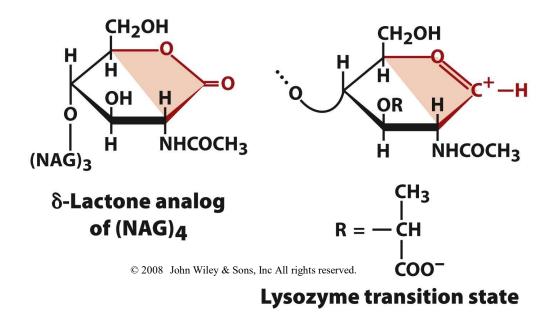
# **Lysozyme Reaction Mechanism**

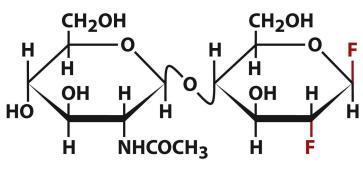


#### CH<sub>2</sub> — CH<sub>2</sub> — Glu35 **Lysozyme Reaction Mechanism** CH<sub>2</sub> -Asp52 **General** base take H<sup>+</sup> © 2008 John Wiley & Sons, Inc All rights reserved. catalysis $CH_2 - CH_2 - Glu35$ $CH_2 - CH_2 - Glu35$ HOCH<sub>2</sub> water HOCH<sub>2</sub> replacing HOCH<sub>2</sub> E ring Oxonium ion (transition state) CH<sub>2</sub> —Asp52 CH<sub>2</sub> —Asp52 CH<sub>2</sub> -CH<sub>2</sub> -Glu35 electrophile attack



## Lysozyme: The Use of Transition State Analog Inhibitor





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**NAG2FGIcF** 

Asp 52

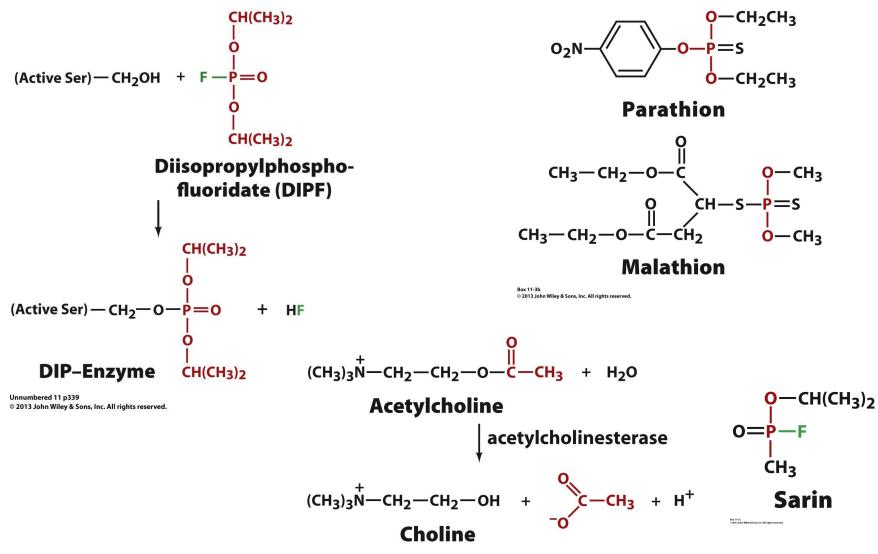
Inhibitor used to verify covalent lysozyme intermediate

#### **Serine Proteases**

- The catalytically active Ser, His, and Asp residues of serine proteases were identified by chemical labeling and structural analysis.
- A binding pocket determines the substrate specificity of the various serine proteases.
- Serine proteases catalyze peptide bond hydrolysis via proximity and orientation effects, acid—base catalysis, covalent catalysis, electrostatic catalysis, and transition state stabilization.
- Zymogens are the inactive precursors of enzymes.

## **DIPF Irreversibly Inactivates Serine Proteases**

: nerve gases



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# **Affinity Labeling: Trypsin & Chymotrypsin**

Trypsin

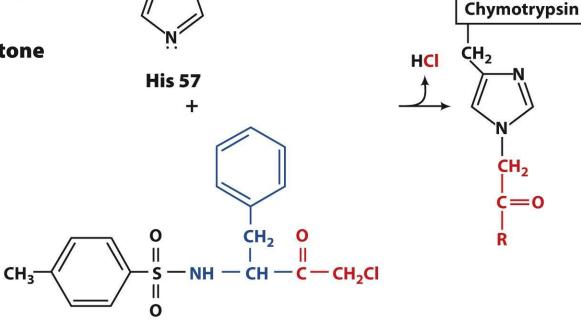
$$CH_2$$
 $CH_2$ 
 $CH_2$ 
 $CH_2$ 
 $CH_2$ 
 $CH_2$ 
 $CH_2$ 
 $CH_3$ 
 $CH_2$ 
 $CH_3$ 
 $CH_$ 

**Tosyl-L-lysine chloromethylketone** 

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#### Chymotrypsin

c=0



Tosyl-L-phenylalanine chloromethylketone (TPCK)

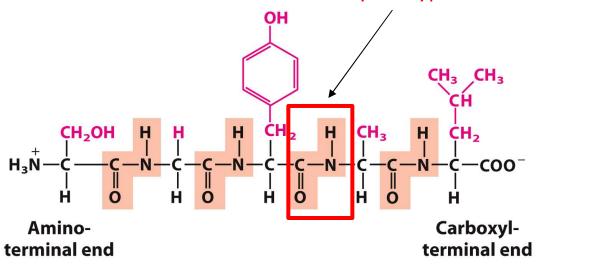
Chymotrypsin

CH<sub>2</sub>

## Chymotrypsin

- During digestion, dietary proteins must be broken down into small peptides by proteases.
- Chymotrypsin is one of several proteases that cuts peptides at specific locations on the peptide backbone.
- This protease is able to cleave the peptide bond adjacent to aromatic amino acids.

Chymotrypsin cuts this bond.



Lehninger Principles of Biochemistry, Seventh Edition

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# **Specificity Pockets of Serine Proteases**

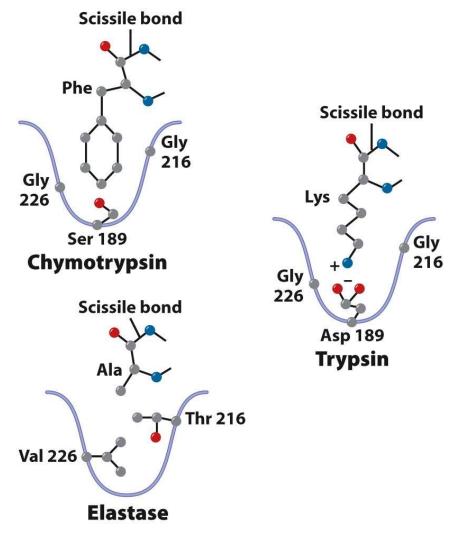


Figure 11-27
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## **Active Site Residues in Serine Proteases**

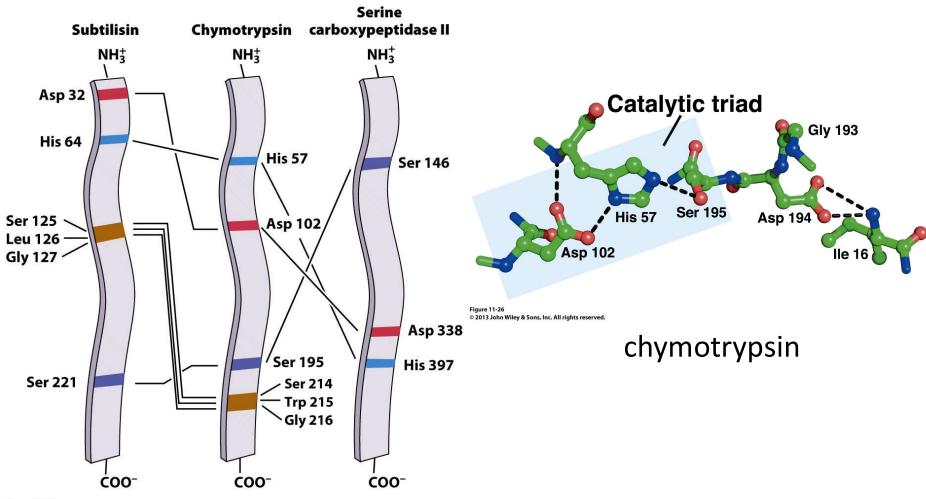
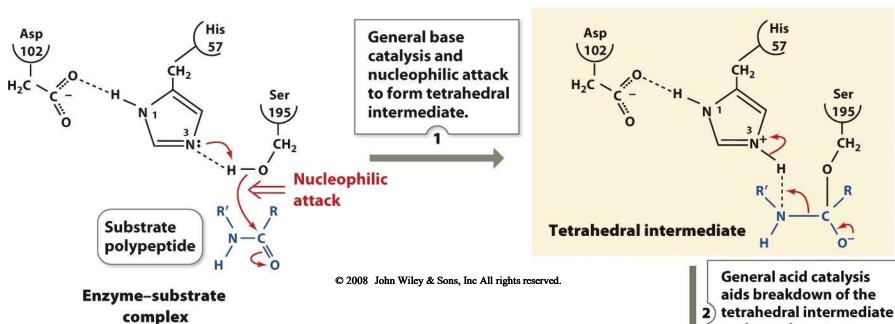
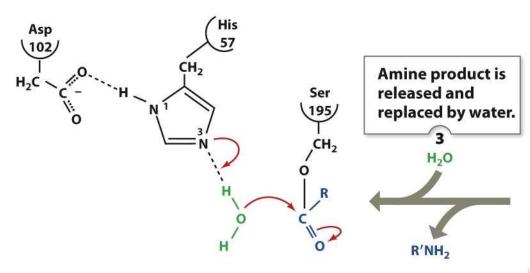


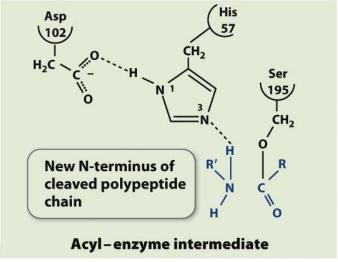
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#### **Mechanism of Serine Proteases**

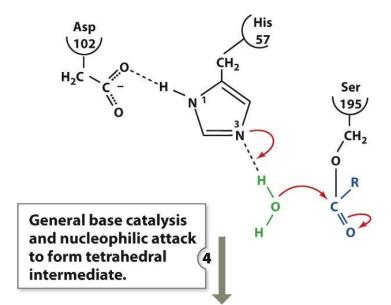




2) tetrahedral intermediate to the acyl-enzyme intermediate.



#### **Mechanism of Serine Proteases**



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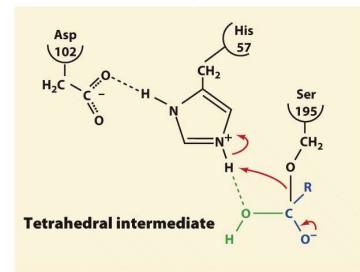
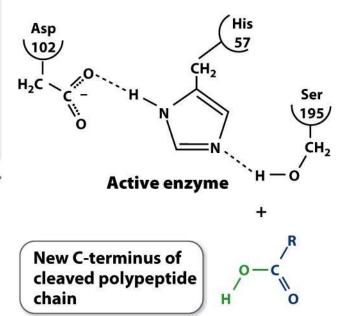
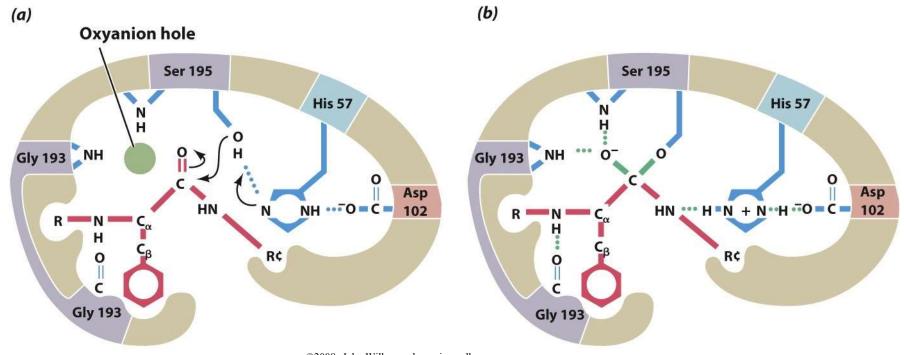


Figure 11-29 part 4 © 2013 John Wiley & Sons, Inc. All rights reserved. General acid catalysis aids breakdown of tetrahedral intermediate to the carboxyl product and the active enzyme.

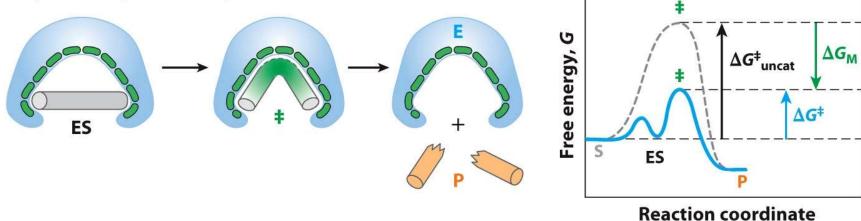


## **TS Stabilization in Serine Proteases**

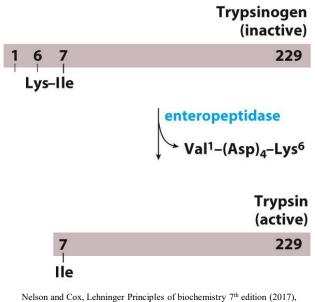


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#### **Enzyme complementary to transition state**



# Zymogens are activated by irreversible covalent modification: blood coagulation cascade



Nelson and Cox, Lehninger Principles of biochemistry 7<sup>th</sup> edition (2017), International edition Figure 6-39 (p232)

**EXTRINSIC PATHWAY** Vascular injury **INTRINSIC PATHWAY Tissue factor** Factor XI Factor XIa Factor VII<sub>a</sub>-Tissue factor Factor VII-**Tissue factor Factor IX** Factor IX Factor IXa VIII **Factor X** Factor Xa Factor X Prothrombin Prothr Thrombin Factor XIII **Fibrin Fibrinogen** Factor XIIIa Cross-linked fibrin

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# **Specificity Pockets of Serine Proteases**

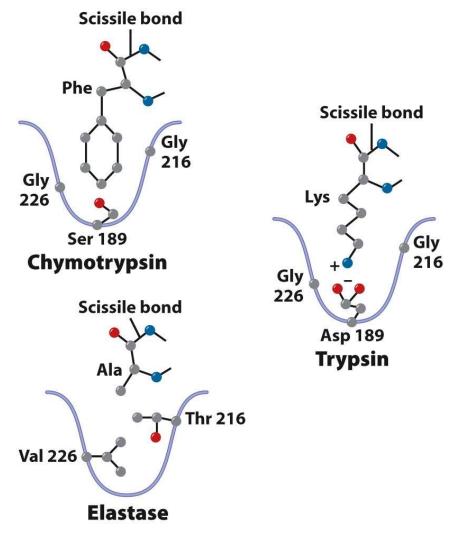


Figure 11-27
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