

Dominant Spectra of Background Radiation in an SF₆ Post-Arc Channel

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Abstract—Composition of background spectra radiated from an SF₆ post-arc channel after current zero was investigated. The radiation intensity of background spectra at a wavelength of 455 nm around current zero was measured. Post-arc temperature was estimated from the intensity ratio of two spectral lines at 426.0 and 442.7 nm of iron used as an electrode material. Emission coefficients of not only continuous spectra but also S₂ spectra due to $B^3\Sigma_u^- - X^3\Sigma_g^-$ transition were theoretically calculated as a function of temperature in the range of 3000–6000 K. The results indicated that the background spectra mainly consist of continuous spectra at temperatures above 4500 K, and of S₂ spectra at temperatures below 4500 K. The temperature dependence of the calculated emission coefficients of the background spectra agreed fairly well with that of the measured radiation intensity.

Index Terms—Background radiation, continuous spectra, S₂ spectra, SF₆ post-arc channel.

I. INTRODUCTION

IN an SF₆ gas-blast circuit breaker, an arc discharge was established between the electrodes during current interruption processes. In order to elucidate the arc interruption phenomena, diagnosis of the arc temperature T and concentration X_M of metallic vapor ejected from the electrodes into the arc is of great importance. In measuring T and X_M in the arc column, spectroscopic observation is often performed to observe not only objective spectral lines but also background spectra near these spectral lines. Thus, it is necessary to investigate the background spectra. The background spectra at $T > 6000$ K is usually considered to be continuous spectra. Until now, several studies have been done to find out various properties of the continuous spectra at $T > 6000$ K [1], [2].

On the other hand, temperature of post-arc channel after current zero decays to magnitudes below 6000 K. However, very little is known about the continuous spectra at $T < 6000$ K. In addition, there are few studies on the composition of the background spectra at $T < 6000$ K.

The present paper describes the background spectra radiated from an SF₆ post-arc channel. First, time variation of radiation intensity I_{455} of background spectra at a wavelength of 455 nm was measured around current zero in a flat-type SF₆ gas-blast

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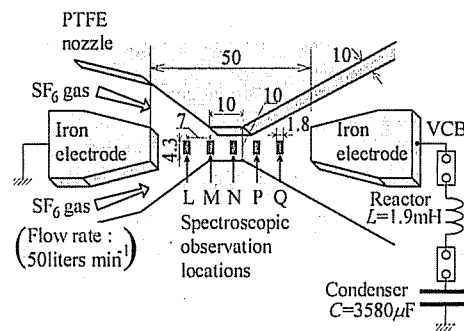


Fig. 1. Flat-type SF₆ gas-blast quenching chamber and spectroscopic observation position.

quenching chamber. In the chamber, we intentionally used iron electrodes, and observed Fe spectral lines at wavelengths of 426.0 and 442.7 nm together with the background spectra. This is because these Fe spectral lines have much higher radiation intensities than those of S⁺, F, or Cu at $T < 6000$ K [3], [4]. Application of the two-line method to the measured two Fe spectral lines enabled us to estimate the post-arc temperature T up to 100 μ s after current zero for ac 5 kA_{peak} arcs [3], [4]. As a result, the radiation intensity I_{455} of the observed background spectra was successfully obtained as a function of T . Second, to interpret the temperature dependence of I_{455} , integrated emission coefficient ϵ'_{cont} of the continuous spectra at 455 nm was theoretically calculated as a function of T in the range from 3000 to 6000 K. However, the temperature dependence of ϵ'_{cont} shows no agreement with that of the observed I_{455} . Finally, special attention was given to molecular spectra, where the integrated emission coefficient $\epsilon_{S_2,455}$ of S₂ spectra due to $B^3\Sigma_u^- - X^3\Sigma_g^-$ transition has to be taken into account for the background spectra at $T < 6000$ K, since the temperature dependence of the sum of $\epsilon_{S_2,455}$ and ϵ'_{cont} agrees fairly well with that of I_{455} .

II. MEASUREMENT OF RADIATION INTENSITIES OF IRON SPECTRAL LINES AND BACKGROUND SPECTRA

A. Experimental Setup

Fig. 1 depicts the flat-type SF₆ gas-blast quenching chamber adopted in the experiment [3], [4]. In this chamber, we intentionally used two iron electrodes. A flat nozzle is made of polytetrafluoroethylene (PTFE) with a thickness of 10 mm. The nozzle throat is 10 mm wide and 10 mm long. The flat nozzle was sandwiched between two ceramic plates.

Sulfur hexafluoride gas was supplied through tubes along the cathode to blow in the axial direction to the arc. The flow rate was adjusted to 50 l/min⁻¹ before arc ignition. Gas pressure in the chamber was always kept to be nearly 0.1 MPa.

A damping sinusoidal current with a frequency of 60 Hz was supplied from a capacitor bank (with capacitance C of 3580 μ F) through a reactor (with inductance L of 1.9 mH). Peak value of the current was adjusted to 5000 A. An auxiliary vacuum circuit breaker (VCB) was intentionally connected in series with this flat-type quenching chamber, thus interrupting the sinusoidal current at the first current zero. As a result, no transient recovery voltage (TRV) was applied between the electrodes in the flat-type quenching chamber.

B. Spectroscopic Observation Along Nozzle Axis

Spectroscopic observation was carried out at five positions around the nozzle throat as denoted by the symbol L-Q in Fig. 1. At these positions, the ceramic plate sandwiching the flat nozzle has a rectangular hole of 4 mm \times 1 mm. Through this hole, the arc radiation was observed with a bundle of four optical fibers made of quartz glass. Each fiber has a core of 0.8 mm in diameter, a clad of 1.0 mm in diameter, and a numerical aperture of 0.2. The bundle of fibers accepts arc radiation from a region with spatial dimensions of 4.3 mm \times 1.8 mm.

This bundle of fibers transmits the arc radiation from the observation point to the entrance slit of a monochromator. The monochromator has a reciprocal dispersion of 0.8 nm \cdot mm⁻¹. On the focal plane at the exit of the monochromator, a multichannel detector with three photomultiplier tubes was positioned. Use of this detector permits the simultaneous observation of radiation intensities at three different wavelengths. With the multichannel detector, we measured the radiation intensities I'_{426} and I'_{443} at wavelengths of 426.0 and 442.7 nm, respectively. Simultaneously, the radiation intensity I_{455} at a wavelength of 455.0 nm was also measured to obtain the intensity of background spectra. The wavelength of 455.0 nm is free from the spectral lines emitted from atoms and atomic ions. The observation width of each channel corresponds to 0.6 nm in wavelength. The relative spectral sensitivities of the overall optical system for the above three wavelengths were measured using a standard tungsten-halide lamp. Considering the sensitivities thus obtained, corrections were made to the three signals.

C. Time Variations of Radiation Intensities of Spectra and Temperature around Current Zero

Numerical subtraction of background radiation intensity I_{455} from I'_{426} produces the net radiation intensity I_{Fe426} of Fe spectral line at 426.0 nm ($z^7D^o-e^7D$). Similarly, the net radiation intensity I_{Fe443} of Fe spectral line at 442.7 nm ($a^5D-z^7F^o$) was also derived. The top part of Fig. 2 illustrates the time variations of radiation intensities I_{Fe426} , I_{Fe443} , and I_{455} at position M. As can be seen in this figure, all of I_{Fe426} , I_{Fe443} , and I_{455} were detectable up to 100 μ s after current zero. It should be noted that I_{Fe426} and I_{Fe443} decay rapidly with time, while I_{455} of the background spectra almost

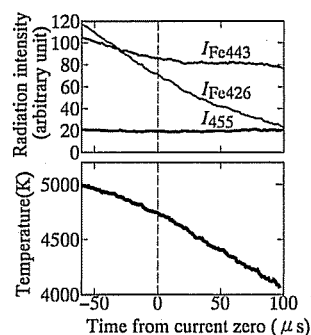


Fig. 2. Time variations of radiation intensity measured around current zero and temperature estimated at position M (Peak value of current, 5 kA; SF₆ gas flow rate, 50 l/min).

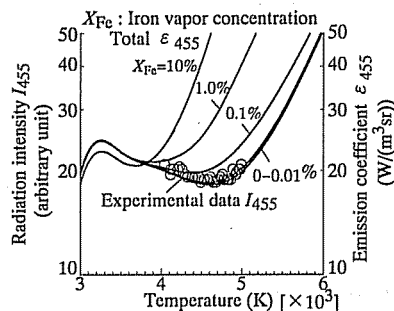


Fig. 3. Temperature dependence of the radiation intensity of background spectra at 455 nm measured at position M and those of emission coefficients calculated.

remain constant. A similar aspect was found at the other observation positions.

From the ratio I_{Fe443}/I_{Fe426} , temperature T around current zero was estimated by the two-line method on the assumption that the population of excited states in Fe follows a Boltzmann distribution and that the arc is optically thin. The validity of these assumptions was confirmed in additional experiments by measuring simultaneously Fe spectral lines at 430.8, 432.6, 438.6, 440.5, and 441.5 nm, as well as at 426.0 and 442.7 nm with an optical multichannel analyzer and by making a Boltzmann plot diagram for the seven lines. The bottom figure in Fig. 2 represents the decay process of T at the position M. Note that T is so-called an excitation temperature of Fe. The temperature T was estimated to be 4870 K at the current zero. It can be also seen that T decreases gradually with time, then reaching to 4140 K at 100 μ s after current zero.

III. TEMPERATURE DEPENDENCE OF RADIATION INTENSITY OF BACKGROUND SPECTRA

Use of Fig. 2 permits us to obtain I_{455} as a function of T . Open circles in Fig. 3 designate I_{455} as a function of T . The intensity I_{455} decreases with T from 5000 to 4600 K, while I_{455} increases slightly with a reduction of T from 4600 to 4100 K. In order to interpret the above relationship between I_{455} and T , emission coefficient ϵ_{455} of background spectra at 455 nm will be discussed in the next section.

IV. THEORETICAL CALCULATION OF EMISSION COEFFICIENTS OF BACKGROUND SPECTRA

A. Equilibrium Composition of SF₆-Fe Mixture Plasma

The calculations of emission coefficients require the particle composition of SF₆-Fe mixture. Thus, we calculated the equilibrium composition of SF₆ gas contaminated with iron vapor [3], [4]. The species taken into account were the molecular, atomic, and ionic species of SF₆, SF₄, SF₂, F₂, S₂, F, S, F⁺, S⁺, F⁻, S⁻, F²⁺, S²⁺, Fe, Fe⁺, Fe²⁺, and electron. We obtained the composition by solving the simultaneous equations: Saha's equation for ionization reactions, Gulberg-Waage's equations for dissociation reactions, the charge neutrality equation, Dalton's laws of partial pressures, and the equation defining the iron vapor concentration X_{Fe} . In the present paper, X_{Fe} (in %) is defined as the ratio of the total mass of Fe, Fe⁺, and Fe²⁺ to the sum of those of all particles:

$$X_{\text{Fe}} = \frac{m_{\text{Fe}}(N_{\text{Fe}} + N_{\text{Fe}^+} + N_{\text{Fe}^{2+}})}{\sum_j m_j N_j} \times 100. \quad (1)$$

B. Emission Coefficient of Continuous Spectra

Background spectra are usually considered to be continuous spectra, especially at $T > 6000$ K. Thus, we calculated the emission coefficient of the continuous spectra taking into account four kinds of radiative mechanisms: 1) recombination radiation, 2) radiation by electron attachment, 3) bremsstrahlung, and 4) radiation by collision between electron and neutral species [5].

1) *Recombination Radiation*: The emission coefficient ϵ_{fb} (in $\text{W} \cdot \text{m}^{-4} \cdot \text{sr}^{-1}$) of the continuous spectra due to recombination at a wavelength of λ is given by

$$\epsilon_{\text{fb}} = C_1 \frac{N_e}{T^{1/2}} \left[1 - \exp\left(-\frac{hc}{kT\lambda}\right) \right] \times \sum_{j(\text{ion})} \frac{g_{1j} Z_{\text{eff}j}^2 N_j}{U_j} \xi_j \frac{c}{\lambda^2} \quad (2)$$

$$C_1 = \frac{1}{(4\pi\epsilon_0)^3} \frac{16\pi e^6}{3c^3 (6\pi m_e^3 k)^{1/2}}$$

where e the electronic charge, m_e the mass of electron, h Planck's constant, c the velocity of light, k Boltzmann's constant, g_{1j} the statistical weight of the ground state of species j , and U_j the internal partition function of species j . The quantities N_e and N_j are the number density of the electron and the species j , respectively, $Z_{\text{eff}j}$ is the effective ionic charge of the species j , and ξ_j is the Bibermann-Schlüter factor for nonhydrogenic particles [6]. The wavelength λ was taken to be 455.0 nm.

2) *Radiation by Electron Attachment*: Atoms such as F and S can capture a free electron to form negative ions F⁻ and S⁻ with the emission of a continuum similar to that of the recombination process. However, the attachment continuum of F⁻ occurs at $\lambda < 364.6$ nm because of the existence of the absorption edges [7], while that of S⁻ at $\lambda < 614.4$ nm [8]. Thus, the electron attachment to S mainly contributes to the

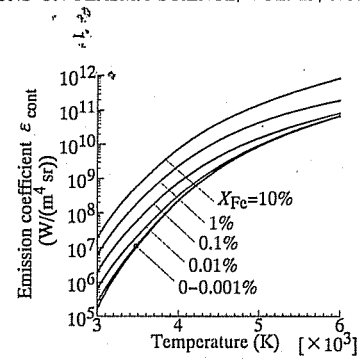


Fig. 4. Emission coefficient ϵ_{cont} of continuous spectra at a wavelength of 455.0 nm in SF₆ gas at 0.1 MPa as a function of temperature.

light emission at 455.0 nm. The emission coefficient ϵ_{attach} of the light at λ is expressed as

$$\epsilon_{\text{attach}} = \frac{2hc}{\lambda^3} \exp\left(-\frac{hc}{kT\lambda}\right) N_{\text{S}^-} \sigma_{\text{det}}^{\text{S}^-} \frac{c}{\lambda^2} \quad (3)$$

where N_{S^-} is the number density of S⁻, $\sigma_{\text{det}}^{\text{S}^-}$ is the photo-detachment cross sections for S⁻.

3) *Bremsstrahlung*: The emission coefficient ϵ_{ff} of the continuous spectra due to free-free transitions of electrons can be calculated by the following equation:

$$\epsilon_{\text{ff}} = C_1 \frac{N_e}{T^{1/2}} \exp\left(-\frac{hc}{kT\lambda}\right) \times \sum_{j(\text{ion})} Z_{\text{eff}j}^2 N_j G_{\text{ff}} \frac{c}{\lambda^2} \quad (4)$$

where G_{ff} is the temperature-averaged free-free Gaunt factor, which is close to unity [9].

4) *Radiation by Collision Between Electron and Neutral Species*: The emission coefficient ϵ_{en} of continuum due to collisions between electron and neutral particles is given as follows [5]:

$$\epsilon_{\text{en}} = \frac{32e^2}{3c^3 (4\pi\epsilon_0)} \left(\frac{kT}{2\pi m_e}\right)^{3/2} N_e \exp\left(-\frac{hc}{kT\lambda}\right) \times \sum_{j(\text{neutral})} N_j G_{\text{nj}} \frac{c}{\lambda^2} \quad (5)$$

where G_{nj} is a factor depending on the cross sections of elastic electron-neutral collision.

The calculation results revealed that ϵ_{attach} is very much higher than ϵ_{fb} , ϵ_{ff} , and ϵ_{en} in the temperature range from 3000 to 6000 K. For example, at $T = 5000$ K, ϵ_{attach} is more than 130 times greater than those by the other three processes.

The summation of ϵ_{fb} , ϵ_{attach} , ϵ_{ff} , and ϵ_{en} yields the total emission coefficient ϵ_{cont} of the continuous spectra

$$\epsilon_{\text{cont}} = \epsilon_{\text{fb}} + \epsilon_{\text{attach}} + \epsilon_{\text{ff}} + \epsilon_{\text{en}}. \quad (6)$$

Fig. 4 indicates ϵ_{cont} as a function of T in the range of 3000–6000 K at $P = 0.1$ MPa with a parameter of X_{Fe} . As seen in this figure, ϵ_{cont} decreases markedly with reducing T for the same X_{Fe} .

The measured I_{455} included radiation intensities integrated in wavelength range of 455.0 ± 0.3 nm. Thus, we calculated

the integrated emission coefficient ϵ'_{cont} (in $\text{W} \cdot \text{m}^{-3} \cdot \text{sr}^{-1}$) of continuum in the range of 455.0 ± 0.3 nm by

$$\epsilon'_{\text{cont}} = \int_{454.7 \text{ nm}}^{455.3 \text{ nm}} \epsilon_{\text{cont}} d\lambda \simeq \epsilon_{\text{cont}} \times (0.6 \text{ nm}). \quad (7)$$

Although not being able to compare values of I_{455} with the absolute values of ϵ'_{cont} , we are able to compare the temperature dependence of I_{455} with that of ϵ'_{cont} , i.e., that of ϵ_{cont} . It should be noted that the temperature dependence of ϵ_{cont} greatly differs from that of I_{455} shown with the open circles in Fig. 3. This result indicates that background spectra include significant spectra besides the continuous spectra in the temperature range of 3000–6000 K.

C. Emission Coefficient of S₂ Spectra Due to B³Σ_u⁻–X³Σ_g⁻ Transition

In the temperature range of 3000–6000 K, the high-temperature SF₆ gas has the molecules such as S₂, F₂, and SF₂. Thus, particular attention to the molecular spectra was paid especially to S₂ spectra. The reasons are given below. 1) SF₂ has much lower number density than S₂. 2) F₂ emits band spectra with marked low intensity, because F₂ has much higher electronic energy of upper state than S₂ [10].

Sulfur molecules S₂ emit very extensive band spectra mainly through B³Σ_u⁻–X³Σ_g⁻ transition [11]. The emission coefficient ϵ_{S_2} (in $\text{W} \cdot \text{m}^{-3} \cdot \text{sr}^{-1}$) for B³Σ_u⁻–X³Σ_g⁻ transition of S₂ is written by

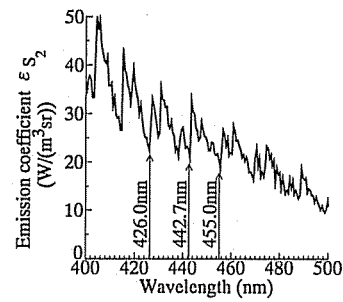
$$\epsilon_{\text{S}_2} = \frac{hc}{4\pi \lambda_{v',v'',J',J''}} N_{n',v',J'} A_{v',v'',J',J''} \quad (8)$$

where n, v, J are the quantum numbers on electronic, vibrational, and rotational state, respectively. The prime (') denotes the upper state, and the double prime (')' the lower state. Then, $\lambda_{v',v'',J',J''}$ is the wavelength, and $A_{v',v'',J',J''}$ the transition probability. The calculation was performed for vibrational transitions $(v', v'') = (0, 0), (0, 1), \dots, (12, 25), (12, 26)$ and for rotational transitions including P and R branches with $J' = 0, \dots, 500$. Fig. 5 represents ϵ_{S_2} at $P = 0.1$ MPa as a function of wavelength at $T = 4000$ K. This figure reveals that S₂ has differences of 9% in emission coefficients at the wavelengths of 426.0, 442.7, and 455.0 nm. This difference may cause an error in temperature estimates. However, the error is only 20 K at current zero in Fig. 2 for example. Thus, emission coefficient of S₂ at 455.0 nm can be regarded as background spectra at the three wavelengths.

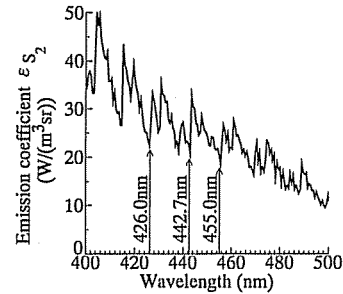
The integrated emission coefficient $\epsilon_{\text{S}_2,455}$ in the wavelength range of 455.0 ± 0.3 nm was expressed as follows:

$$\epsilon_{\text{S}_2,455} = \sum_{455.0 \pm 0.3 \text{ nm}} \epsilon_{\text{S}_2}. \quad (9)$$

Fig. 6 represents $\epsilon_{\text{S}_2,455}$ as a function of T for X_{Fe} from 0 to 10%. As can be seen in this figure, $\epsilon_{\text{S}_2,455}$ increases gradually with reducing T . For example, $\epsilon_{\text{S}_2,455}$ has the magnitude of $10.8 \text{ W} \cdot \text{m}^{-3} \cdot \text{sr}^{-1}$ at 6000 K, while $\epsilon_{\text{S}_2,455}$ is $19.8 \text{ W} \cdot \text{m}^{-3} \cdot \text{sr}^{-1}$ at 3000 K. From Fig. 6, $\epsilon_{\text{S}_2,455}$ is also found to be almost independent of X_{Fe} in range of 0–1% for a given T .



(a)



(b)

Fig. 5. Dependence of emission coefficient due to S₂ B³Σ_u⁻–X³Σ_g⁻ transition on wavelength in pure SF₆ gas at 4000 K at 0.1 MPa.

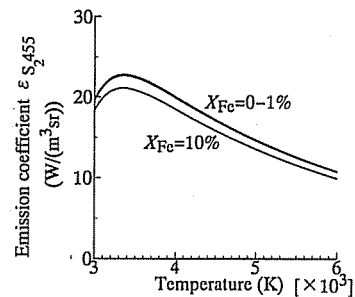


Fig. 6. Emission coefficient due to S₂ B³Σ_u⁻–X³Σ_g⁻ transition in wavelength range from 454.7 to 455.3 nm in SF₆ gas at 0.1 MPa as a function of temperature.

V. COMPARISON OF CALCULATED EMISSION COEFFICIENT WITH MEASURED RADIATION INTENSITY OF BACKGROUND SPECTRA

Let us calculate total integrated emission coefficient ϵ_{455} of background spectra in the range of 455.0 ± 0.3 nm:

$$\epsilon_{455} = \epsilon'_{\text{cont}} + \epsilon_{\text{S}_2,455}. \quad (10)$$

In Fig. 3, bold curves represent ϵ_{455} as a function of T at $P = 0.1$ MPa. It should be noted that the temperature dependence of ϵ_{455} for $X_{\text{Fe}} = 0$ –0.01% agrees fairly well with that of I_{455} . This agreement indicates that the background spectra at the wavelength of 455 nm mainly consist of not continuous spectra but S₂S₂ spectra through B³Σ_u⁻–X³Σ_g⁻ transition at $T < 4500$ K. In addition, the agreement also suggests that X_{Fe} in the post-arc channel was probably of order of 0–0.01%.

VI. CONCLUSIONS

We investigated background spectra in an SF₆ post-arc channel after current zero. Spectroscopic observations were performed to obtain radiation intensity I_{455} of the background spectra at 455 nm as a function of temperature T . The intensity I_{455} proved to decrease with T from 5000 to 4600 K, while to increase slightly with T from 4600 to 4100 K. To interpret the aspect of this temperature dependence, we first calculated emission coefficient ϵ_{cont} of continuous spectra in the temperature range of 3000–6000 K. However, the temperature dependence of ϵ_{cont} shows no agreement with that of I_{455} in the temperature range from 3000 to 5000 K. The emission coefficient $\epsilon_{S_2,455}$ of S₂ spectra due to $B^3\Sigma_u^- - X^3\Sigma_g^-$ transition at 455 nm were theoretically calculated as a function of T . Then, we derived the emission coefficient ϵ_{455} of background spectra by summing $\epsilon_{S_2,455}$ and ϵ'_{cont} at 455 nm. The temperature dependence of ϵ_{455} agreed fairly well with that of I_{455} . This result indicates that the background spectra mainly consist of continuous spectra at $T > 4500$ K, while of S₂ spectra at $T < 4500$ K.

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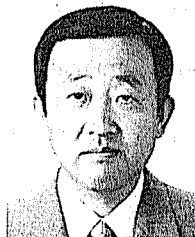
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