

# Effect of nitrogen atoms included in CN<sub>x</sub> coatings on friction sliding against Si<sub>3</sub>N<sub>4</sub> ball in nitrogen gas

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## Abstract:

Carbon nitride (CN<sub>x</sub>) coatings are thought to have good tribological properties because of low friction and high wear resistance. This material shows a friction coefficient lower than 0.01 when slid against a Si<sub>3</sub>N<sub>4</sub> ball in dry N<sub>2</sub> gas. The mechanism of superlow friction was believed to be due to the change in the topmost layer of the CN<sub>x</sub> surface in the wear track to a low-shear-strength layer by forming a graphite-like structure. However, the effects of nitrogen atoms in the CN<sub>x</sub> coating and the ambient N<sub>2</sub> gas on the transformation of the surface layers of CN<sub>x</sub> remain unclear.

In this work, we investigated the relationship between the friction coefficient and the nitrogen concentration in the wear track on the CN<sub>x</sub> coating. We measured the intensity of C KLL and N KLL on the wear track of CN<sub>x</sub> by Auger electron spectroscopy (AES), and conducted Raman analysis. We compared the N KLL/C KLL ratios of wear tracks from sliding in ambient N<sub>2</sub>, Ar, and O<sub>2</sub>. Only in the case of the wear track from material slid in N<sub>2</sub>, did the N KLL/C KLL ratio decrease to almost zero and the friction coefficient became lower than 0.01. On the other hand, in the Ar environment, the N KLL/C KLL ratio in the wear track was comparable to the as-deposited CN<sub>x</sub> coating, the friction coefficient became 0.03, and the I<sub>G</sub> position of Raman shift increased at most 3 cm<sup>-1</sup>. The N KLL/C KLL intensity of the wear track after friction tests in Ar and then in N<sub>2</sub> was approximately 0.2 times as low as the as-deposited CN<sub>x</sub>. The friction coefficient in N<sub>2</sub> gas was 0.03 when incorporated nitrogen remained in the CN<sub>x</sub> surface. On the other hand, the friction coefficient was lower than 0.03 when incorporated nitrogen was desorbed.

## 1. INTRODUCTION

Carbon nitride is a material considered to be harder than diamond and was reported about 20 years ago [1, 2]. There is no perfect crystal of bulk β-C<sub>3</sub>N<sub>4</sub>, however, some researchers investigated the relationship between the coating method and the CN<sub>x</sub> structure [3-5], the optical properties [6-9], and the hardness or proof of wear [10-13].

Umehara et al. reported a superlow friction coefficient lower than 0.01 without a lubricant which had been obtained when amorphous CN<sub>x</sub> coating slid against a Si<sub>3</sub>N<sub>4</sub> ball after several 10-10<sup>3</sup> cycles in dry nitrogen gas (the so-called "running-in" period). The mechanism of the low friction coefficient was considered to be the sliding surface of CN<sub>x</sub> changing to a graphite-like structure [14, 15]. Furthermore,

AES (Auger Electron Spectroscopy) analysis revealed that nitrogen atoms were desorbed from the CN<sub>x</sub> wear surface [16], and the shearing strength and ploughing strength values became lower than as-deposited CN<sub>x</sub> [17]. As mentioned above, we suggested that nitrogen atoms incorporated in the CN<sub>x</sub> coating generate a friction force between this surface and a mating material.

First, we carried out AES depth analysis for the distribution of nitrogen concentration in order to clarify the deposition procedure which would produce superlow friction surface during the running-in period. Second, in order to clarify the effect of nitrogen gas on desorption from the CN<sub>x</sub> coating, we carried out AES and Raman analysis to compare the nitrogen concentration of each of the wear tracks from sliding in N<sub>2</sub>, Ar, and O<sub>2</sub> gas. Finally, we analyzed the CN<sub>x</sub> wear track from sliding in Ar gas until the friction coefficient reached about 0.03, after which that the same wear track was used for sliding in N<sub>2</sub> gas. We concluded that nitrogen atoms did not desorb from the CN<sub>x</sub> coating when a friction test was performed in Ar gas. Thus, we clarified the effect of nitrogen atom incorporation in the CN<sub>x</sub> coating surface on the coefficient of friction.

## 2. MATERIALS AND EXPERIMENTAL

CN<sub>x</sub> coatings were deposited 100 nm thick on Si(100) substrates by the Ion Beam assisted Mixing method [18, 19]. The CN<sub>x</sub> coating deposition conditions were summarized in Table I [20]. Umehara et al. reported that consistent values of 398.5 and 399.5 eV for N1s were observed from XPS; the I<sub>G</sub> peak position was 1549 cm<sup>-1</sup>, and the I<sub>D</sub>/I<sub>G</sub> ratio was 1.0 [15]. Therefore, we concluded this CN<sub>x</sub> coating had 0-20 % sp<sup>3</sup> carbon bonds [21]. The mating part in the friction test was a Si<sub>3</sub>N<sub>4</sub> ball with a diameter of 8 mm. The hardness and Young's modulus of CN<sub>x</sub> and a Si<sub>3</sub>N<sub>4</sub> ball were summarized in Table II.

A friction test apparatus was placed inside a vacuum chamber [22]. Before the friction tests, we evacuated this chamber by rotary pump to 0.1 Pa or lower. The chamber was filled for different tests with N<sub>2</sub> (99.998% purity), Ar (99.998% purity), and O<sub>2</sub> (99.998% purity) gas at atmospheric pressure. We started the friction test after the O<sub>2</sub> concentration in the chamber became less than 100 ppm when measuring friction in N<sub>2</sub> and in Ar. For tests in O<sub>2</sub>, we

Table CNx coating deposition conditions

|  |                      |
|--|----------------------|
| Substrate                              | Si(100)              |
| Background pressure (Pa)               | $1.0 \times 10^{-4}$ |
| Sputter-cleaning (min)                 | 5.0                  |
| Sputter-cleaning acc. voltage (kV)     | 1.0                  |
| Sputter-cleaning acc.curent (mA)       | 0.1                  |
| Deposition conditions                  |                      |
| Carbon target (99.999% purity)         |                      |
| Sputtering ion beam (99.99998% purity) | Ar                   |
| Acc. voltage (kV)                      | 1.0                  |
| Acc. current (mA)                      | 100                  |
| Nitrogen mixing conditions             |                      |
| Sputtering ion beam (99.99998% purity) |                      |
| Acc. voltage (kV)                      | 0.5                  |
| Acc. current (mA/cm <sup>2</sup> )     | 0.03                 |

started it after the O<sub>2</sub> concentration was more than 99.99%. Friction force and normal load were measured by strain gauges posted on leaf springs. The sensitivities were 0.0265 mN/mV for the friction force and 0.047 mN/mV for the normal load.

Friction tests were carried out under 0.1 N of normal load with a sliding velocity of 0.01 m/s. For the first section, we carried out 3 friction tests each using N<sub>2</sub>, Ar, and O<sub>2</sub> gases until the number of sliding cycles reached 16000 cycles. For the second section, we carried out a friction test in Ar until the friction coefficient became 0.03, the chamber was then evacuated to 0.1 Pa or lower and filled with N<sub>2</sub> gas, after which we continued friction tests in the same wear track.

We analyzed each wear track by AES (ULVAC PHI-650). In the normal load condition, the calculated Hertzian contact diameter was approx. 20 μm, which was much larger than the AES beam diameter (at most 1 μm). Depth analysis revealed that the acceleration voltage of the Ar<sup>+</sup> beam of the sputter ion gun was 3.0 kV [raster size (about 3 × 3 mm<sup>2</sup>) and the Ar pressure in the ionization chamber was adjusted for 14.4 nm/min on the reference CNx (100 nm)]. We normalized the obtained spectra of C KLL peak-to-background intensity as 1.0.

The Raman spectra were obtained using a Jasco Laser Raman spectrophotometer NRS-1000 equipped with second harmonic of a Nd:YAG laser (532 nm). We obtained the I<sub>D</sub> and I<sub>G</sub> position, and I<sub>D</sub>/I<sub>G</sub> ratio after the deconvolution of Raman spectra.

### 3. EXPERIMENTAL RESULTS

#### 3.1 Friction test in N<sub>2</sub>, Ar, and O<sub>2</sub> and AES analysis for wear track

Figures 1(a)-(c) show the friction behavior with the number of sliding cycles in N<sub>2</sub>, Ar, and O<sub>2</sub> gas. Friction coefficient increased to about 0.4 at the initial stage in the case of friction in N<sub>2</sub> and in Ar. Then, in N<sub>2</sub> condition, the friction coefficient decreased to 0.01 at the end of the given number of sliding cycles, and it reached the steady state. In the friction test in Ar, the friction coefficient decreased to approx. 0.05 rapidly, then gradually decreased to 0.03 with the number of sliding cycles. On the other hand, the friction test with O<sub>2</sub> showed a friction coefficient with an unsteady state. The values varied from 0.1 to 0.8.

Figure 2 shows the AFM images of wear tracks on CNx coating in each of the N<sub>2</sub>, Ar, and O<sub>2</sub> environments. Particularly, CNx coating evidenced little wear in the N<sub>2</sub> and Ar environment. On the other hand, the CNx coating wore to approx. 30 nm depth in the O<sub>2</sub> environment. Figure 3 shows the results of AES analysis of each wear track. The N KLL intensity of as-deposit CNx was approx. 0.2. The N KLL intensity of friction in N<sub>2</sub> was below the detection level, then became almost 0. In the friction test in Ar, the N KLL intensity was approx. 0.2, which was comparable to the as-deposit CNx. On the other hand, the N KLL intensity of friction in O<sub>2</sub> became approx. 0.29, which was 1.5 times larger than as-deposit CNx. Furthermore, it was suggested that the O<sub>2</sub> molecules were adsorbed on the wear track during transport to the AES analysis chamber [20].

The depth distribution of N1s peak-background intensities is shown in Fig. 4. The N KLL intensity of wear track in N<sub>2</sub> gradually increased so as to be comparable to the as-deposit CNx, becoming constant at 10 nm in the depth direction. However, Ar and O<sub>2</sub> results showed that the N KLL intensity distribution was almost constant in the depth direction of coating as it was comparable to the as-deposit CNx without the topmost layer of wear track in O<sub>2</sub>.

To clarify the structural change of CNx surface in Ar and N<sub>2</sub> environment, we analyzed its surfaces with Raman as shown in Fig. 5 (a)-(c). The I<sub>G</sub> peak, and the I<sub>D</sub>/I<sub>G</sub> ratio of the CNx surface slid in N<sub>2</sub> increased from 1556 to 1566 cm<sup>-1</sup>, and from 1.10 to 1.13. However, in the friction test in Ar, the I<sub>G</sub> peak increased from 1556 to 1559 cm<sup>-1</sup>, which

#### 3.2 Effect of incorporation of nitrogen atoms on friction coefficient

Table Indentation hardness and Young's modulus of as-deposit CNx and a Si<sub>3</sub>N<sub>4</sub> ball

| Nanoindentation hardness | Indentation depth (nm) | Indentation hardness |                                     | Young's modulus (GPa) |                                     |
|--------------------------|------------------------|----------------------|-------------------------------------|-----------------------|-------------------------------------|
|                          |                        | CNx                  | Si <sub>3</sub> N <sub>4</sub> ball | CNx                   | Si <sub>3</sub> N <sub>4</sub> ball |
| ENT-1100a                | 20.0                   | 25.2                 | 15.0                                | 467.6                 | 315.0                               |

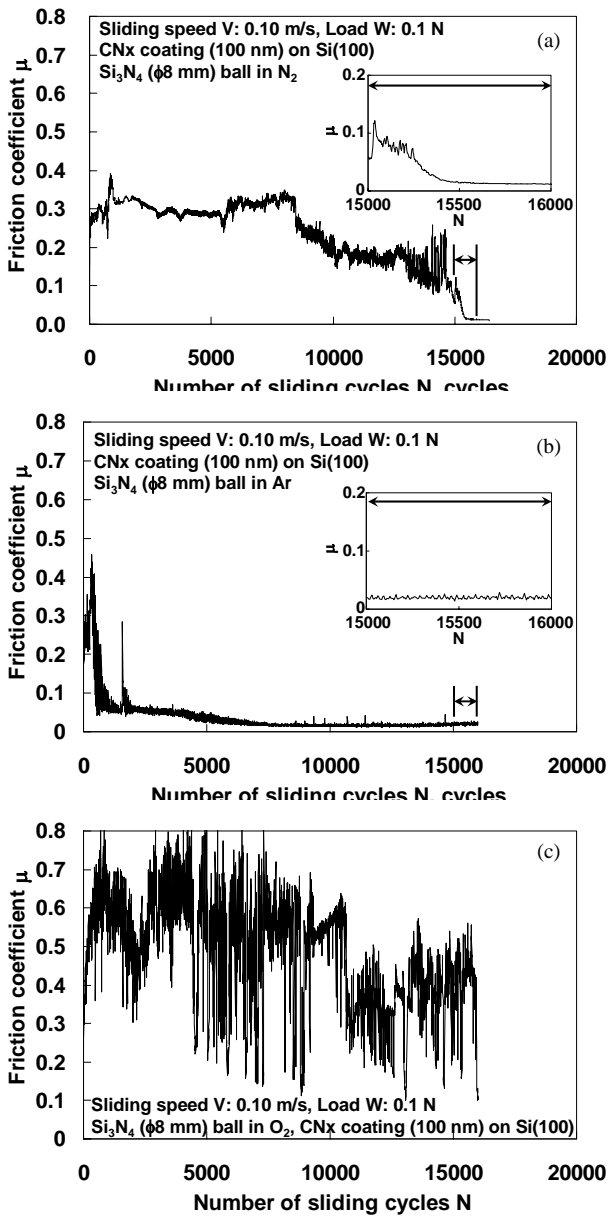


Figure 1 The friction coefficient behavior of each environmental gas (a) in  $\text{N}_2$ , (b) in Ar, and (c) in  $\text{O}_2$ .

Figure 6 (a) shows the friction coefficient behavior of CNx coating sliding against a  $\text{Si}_3\text{N}_4$  ball in  $\text{N}_2$  gas after the friction coefficient became 0.03 in Ar. At the initial stage of the sliding in Ar, the friction test was run until the friction coefficient became approx. 0.03. After that, in the middle stage, the friction test was stopped while the vacuum chamber was evacuated and filled with  $\text{N}_2$  gas at atmospheric pressure. At the end stage, we restarted the friction test at the same wear track in  $\text{N}_2$  gas. The friction coefficient increased to 0.13 just after the replacement of Ar gas with  $\text{N}_2$  gas. Then the friction coefficient decreased to

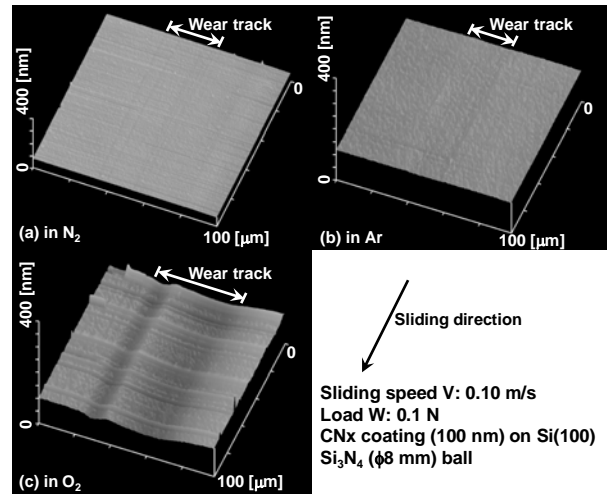


Figure 2 AFM images of wear track on CNx coating, (a) in  $\text{N}_2$ , (b) in Ar, and (c) in  $\text{O}_2$

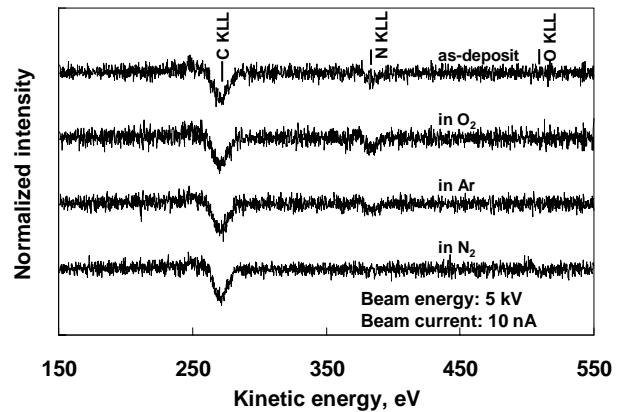


Figure 3 AES analysis for wear track of each environmental gas after 16000 cycles.

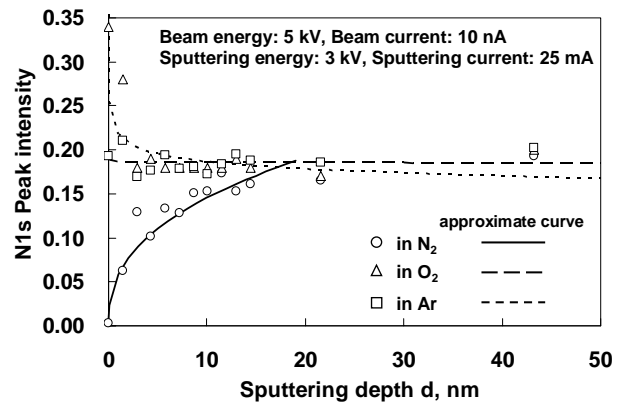


Figure 4 N 1s Peak intensity distribution in CNx coating after friction test in each environmental gas.

lower than 0.01. After the friction test, AES analysis for the wear track was carried out (see Fig. 6 (b)). As mentioned above, the N KLL/C KLL intensity of friction only in Ar gas was comparable to as-deposit CNx. However, the

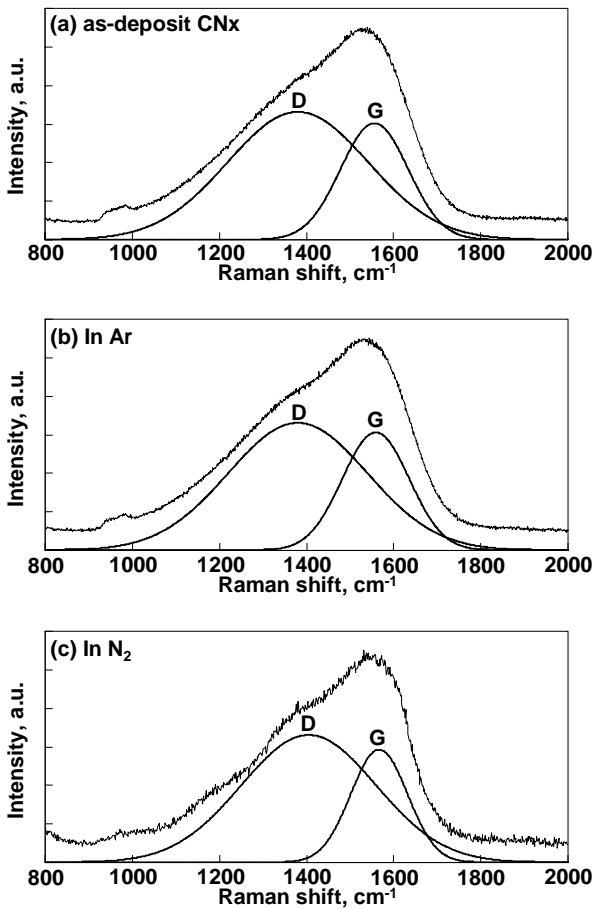


Figure 5 Raman spectra of (a) as-deposit CNx, (b) friction test in Ar, and (c) friction test in N<sub>2</sub>.

intensity of the wear track from sliding in N<sub>2</sub> gas after the friction coefficient became 0.03 in Ar gas decreased to approx. 0.2 times smaller than as-deposit CNx. Therefore, the nitrogen incorporated into the CNx coating surface generated friction force, however, it was effective at a superlow friction coefficient region ( $\mu < 0.03$ ).

#### 4. DISCUSSION

The possibility of nitrogen atom desorption from the CNx coating was considered from the viewpoint of the flash temperature generated by the friction between the CNx coating and the Si<sub>3</sub>N<sub>4</sub> ball. To clarify the effect of the flash temperature on nitrogen desorption, we carried out a 10-min. annealing test for CNx coating at 500°C in N<sub>2</sub>,  $1.0 \times 10^{-3}$  Pa vacuum (HV), and in air. After that, each surface was analyzed by AES as shown in Fig. 7. It was clear that the N KLL/C KLL ratio of annealing in N<sub>2</sub> and in HV decreased from as-deposit CNx. However, the flash temperature was estimated at about 1.4°C [23] at the contact area between the CNx coating and the Si<sub>3</sub>N<sub>4</sub> ball

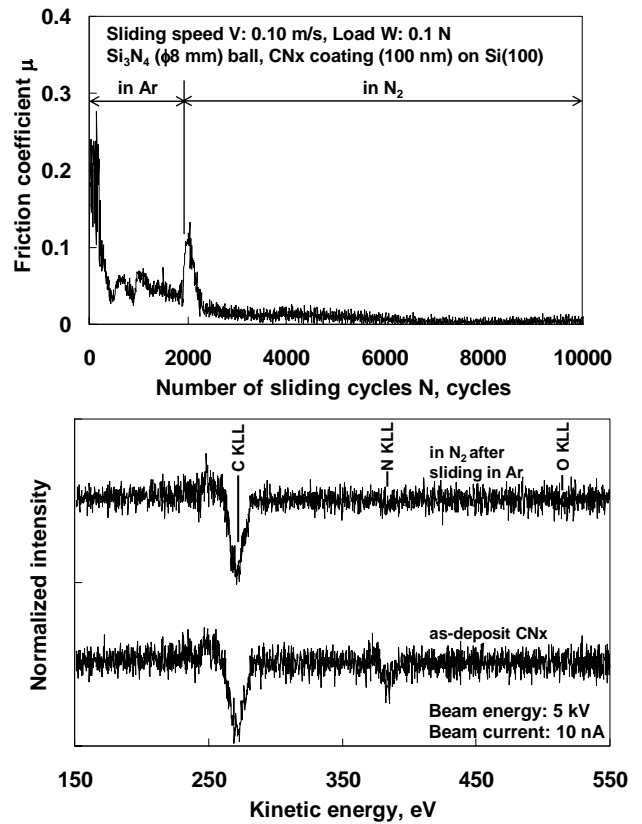


Figure 6 (a) Friction test in N<sub>2</sub> gas from sliding after in Ar, and (b) AES analysis of wear track in N<sub>2</sub> through Ar.

when we assumed that the CNx coating thermal conductivity was 1000-2000 W/m·K [24]. This result indicated that flash temperature generation did not explain the nitrogen non-desorption from CNx slid in Ar because the friction coefficient in Ar gas always has a consistently higher value than in N<sub>2</sub>. Therefore, the flash temperature generated in Ar was higher than in N<sub>2</sub>. On the other hand, in a normal air environment, CNx coating gradually disappeared during annealing. After that it completely disappeared from the Si wafer.

The nitrogen gases were thought to directly affect the nitrogen atom desorption from CNx coating. However, nitrogen molecules are known to be inert in general. Therefore, generation of energy is necessary to break the incorporation between C-N bonds from the friction area in the N<sub>2</sub> environment. Thus, we concluded that it is possible that such energy was generated by tribomicroplasma [25, 26].

The mechanism of nitrogen desorption from the CNx wear track after sliding in N<sub>2</sub> was thought to be the generation of tribomicroplasma. Nakayama et al. reported that tribomicroplasma was generated at the rear port of the sliding contact when a diamond pin was slid against a

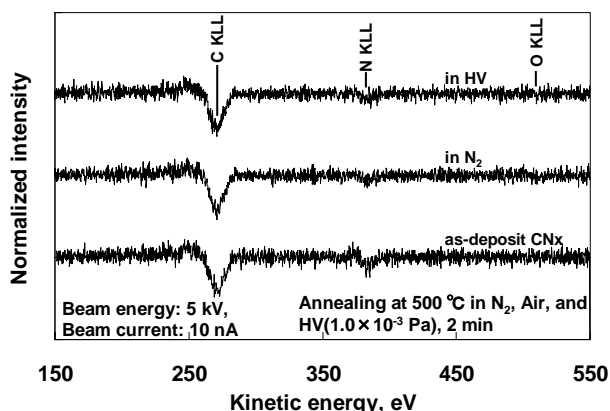


Figure 7 AES analysis for annealing at 500 in HV and in N<sub>2</sub>, and as-deposit CN<sub>x</sub>.

sapphire disk in several environments [25, 26]. Since electrons, ions, and photons were generated from this tribomicroplasma area, several wavelength ultra violet (UV) rays were derived from photons whose generation was related to the surrounding environmental gas. Table III shows several wavelengths of UV generated by tribomicroplasma, the energies calculated from this wavelength, and the relative bonding energies [27]. In N<sub>2</sub> environment, 295-778 nm wavelength-rays were generated. On the other hand, only 696 and 778 nm wavelength-rays were generated in the O<sub>2</sub> environment. Therefore, we concluded that decomposition of C-C and C-N incorporated in CN<sub>x</sub> was more when it was slid in N<sub>2</sub> gas. Then, the decomposed nitrogen atoms were considered to readily become dimmer and desorb from the CN<sub>x</sub> surface [28, 29].

Table The UV ray frequency generated at tribomicroplasma and several bonding energies

| Wave length, nm | energy kJ/mol | Species | Binding energy, kJ/mol |
|-----------------|---------------|---------|------------------------|
|                 |               | H-O     | 460                    |
|                 |               | H-C     | 410                    |
| 295             | 406.11        |         |                        |
|                 |               | H-N     | 390                    |
| 316             | 379.12        |         |                        |
|                 |               | C-O     | 360                    |
| 337             | 355.50        |         |                        |
|                 |               | C-C     | 350                    |
| 358             | 334.64        |         |                        |
| 381             | 314.44        |         |                        |
| 392             | 305.62        |         |                        |
|                 |               | C-N     | 305                    |
| 400             | 299.51        |         |                        |
|                 |               | N-O     | 220                    |
| 696             | 172.13        |         |                        |
|                 |               | N-N     | 160                    |
| 778             | 153.99        |         |                        |
|                 |               | O-O     | 146                    |

## 5. CONCLUSION

In this paper, we first carried out friction tests between CN<sub>x</sub> coating and a Si<sub>3</sub>N<sub>4</sub> ball in N<sub>2</sub>, Ar, and O<sub>2</sub>. We conducted an AES analysis for each wear track to obtain the depth distribution of N KLL intensity. Second, in order to determine the effects of nitrogen atoms incorporated in CN<sub>x</sub> coating on the friction coefficient, we conducted friction tests between CN<sub>x</sub> and a Si<sub>3</sub>N<sub>4</sub> ball in Ar gas until the friction coefficient became 0.03, and then with continued friction tests in N<sub>2</sub> gas after Ar gas was replaced in the friction test chamber. AES analysis was performed on the same wear track in each case. The following results were obtained.

- (1) The N KLL/C KLL intensity of the wear track from sliding in N<sub>2</sub> showed almost 0 at the topmost layer, and N KLL intensity gradually increased to a value comparable to the as-deposit CN<sub>x</sub>. at approx. 10 nm depth.
- (2) There was no nitrogen desorption from CN<sub>x</sub> coating from sliding in Ar gas. The N KLL/C KLL intensity was constant and comparable to the as-deposit CN<sub>x</sub> from the surface to the substrate. Furthermore, the CN<sub>x</sub> coating wore little in Ar.
- (3) The N KLL/C KLL intensity of the wear track was approximately 0.2 times the as-deposit CN<sub>x</sub> whose track was friction-tested in Ar, and then in N<sub>2</sub> after the friction coefficient reached a steady-state value ( $\mu < 0.03$ ) in Ar. The friction coefficient in N<sub>2</sub> gas was 0.03 when the incorporated nitrogen remained in the CN<sub>x</sub> surface. On the other hand, the friction coefficient was lower than 0.03 when the incorporated nitrogen desorbed.

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Carbon nitride (CN<sub>x</sub>) coating shows a friction coefficient lower than 0.01 if it slid against a Si<sub>3</sub>N<sub>4</sub> ball in dry N<sub>2</sub> gas. The mechanism of superlow friction was believed to be due to the change in the topmost layer of the CN<sub>x</sub> surface in the wear track to a low-shear-strength layer by forming a graphite-like structure.

However, the effects of nitrogen atoms in the CN<sub>x</sub> coating and the ambient N<sub>2</sub> gas on the transformation of the surface layers of CN<sub>x</sub> remain unclear. We measured the intensity of C KLL and N KLL on the wear track of CN<sub>x</sub> by AES. The N KLL intensity of as-deposit CN<sub>x</sub> was approx. 0.2. Its intensity of friction in N<sub>2</sub> was lower than can be detected, then it became almost 0. In the case of the friction test in Ar, it was showed that the N KLL intensity was approx. 0.2, which was comparable to the as-deposit CN<sub>x</sub>. On the other hand, the N KLL intensity of friction in O<sub>2</sub> became approx. 0.29 whose value was 1.5 times larger than as-deposit CN<sub>x</sub>. Only in the case of the wear track slid in N<sub>2</sub>, did the N KLL/C KLL ratio decrease to almost zero and the friction coefficient became lower than 0.01, the N KLL intensity gradually increased to a comparable value to the as-deposit CN<sub>x</sub> at approx. 10 nm depths.

